

Rare Gas Solids, Vol. 1 By M. L. Klein;J. A. Venables

By M. L. Klein;J. A. Venables

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A simple equation which predicts the influence of pressure on melting points and includes anharmonicity has been applied to the rare gas solids. L. Klein

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Mechanisms of Melting of Rare Gas Solids 485 this structure differ from the optimal ones by approximately 2.5%. With increasing temperature the amplitude of atom

that are gaseous at room temperature with a density about Kobashi, K.; Klein, M. L. (1980). "Lattice vibrations of solid English, C. A.; Venables, J. A

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LONG and J. SHERIDAN. Pp. viii+ 279. London: The Edited by M. L. KLEIN and J. A. VENABLES. Pp. xiii+ 607. London Crystal and solid state physics

Molecular Photodynamics in Rare Gas Solids V. A. Apkarian* Department of Chemistry, University of California, Irvine, California 92697-2025 N. Schwentner

Additional Physical Format: Online version: Rare gas solids. London ; New York : Academic Press, 1976-1977 (OCoLC)643909094: Document Type: Book: All Authors

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How to Cite (1977), Rare gas solids edited by M. L. Klein and J. A. Venables. Acta Cryst. A, 33: 526. doi: 10.1107/S0567739477001363

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All the rare gases crystallize under sufficient pressure to form close-packed solids that have been extensively studied with the diamond-anvil cell (DAC).

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Thus it is shown that these equations of states successfully explain the behavior of inert gas solids. Documents; M L Klein and J A Venalbes, Rare gas solids

Genre/Form: Resonanzspektroskopie: Additional Physical Format: Online version: Rare gas solids. Berlin ; New York : Springer-Verlag, 1984 (OCoLC)610347295

Desiree, Natural Gas Residual Fluids: Sources, J. L.; Trotz, M.; Venables, R.; Zimmerman, J. B.; Fuchs, V.; Miller, S.; Page, M. J.; Anastas, P. T.;

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